

__E__1(1 point): which of the following will have the greatest atomic radius?

- A. Sodium (Na)
- B. Molybdenum (Mo)
- C. Tungston (W)
- D. Silicon (Si)
- E. Barium (Ba)

__A__2. (1 point) Which of the following will be most likely to lose an electron?

- A. Sodium (Na)
- B. Molybdenum (Mo)
- C. Tungston (W)
- D. Silicon (Si)
- E. Barium (Ba)

Question 3 (2 points): Write the chemical equation for electron affinity for a chlorine atom

